Surface Chemistry for Improvement in Load-Carrying Capacity of Poly(Ether-Ether-Ketone)-Based Materials by Poly(Tetrafluoroethylene)

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The tribological properties of poly(ether-ether-ketone) (PEEK) against steel was evaluated by a ring-on-flat type tribo-test under rotating motion. Addition of poly(tetrafluoroethylene) (PTFE) as solid lubricants improved the load-carrying capacity of the material. The role of PTFE was studied by means of time of flight secondary ion mass spectroscopy (TOF-SIMS). The transfer of PTFE from the PEEK surface onto steel surface was found to be beneficial for preventing seizure.

Keywords: lightweight tribo-material, white metal, Babbit alloy, solid lubricant, PEEK, PTFE, TOF-SIMS, tribo-chemistry

1. Introduction

Babbitt alloys, so-called white metals, are widely employed in bearings as wear resistance tribo-material. Since the operating conditions in bearings are getting harder, materials for those applications have to be improved. Unfortunately, tribological properties of Babbitt alloys are usually insufficient under heavy load conditions. Therefore new materials that possess better tribological properties are desired; we focused on possibility of polymer-based materials. A benefit of the polymer-based materials is possibility of lightweight materials that leads energy saving. Synthetic polymers (density ranging from 1 to 2 g/cm³) are better candidates than lightweight metal alloys (density ranging from 4 to 5 g/cm³). In fact, polymer-based materials have come to apply in mechanical elements such as gears and bearings.

One of candidates for Babbitt alloy alternatives is poly-ether-ether-ketone [PEEK, or formally poly(phenoxyphenylphenylketone)]. They are well known to exhibit good thermal properties. Improvement in mechanical strength of them is achieved by filling carbon fibers1,2). Blending of poly(tetrafluoroethylene) (PTFE) in PEEK have been reported to reduce wear rate and friction3-8) under dry conditions. In this work, the load-carrying capacities of PEEK-PTFE composite were evaluated and the mechanism was investigated with an emphasis on the role of PTFE on the tribological properties from a viewpoint of surface chemistry.

2. Experimental

A thrust type tribo-tester (Figure 1) was employed to evaluate the load-carrying capacity under the conditions listed in Table 1. The composite materials were slid against steel. The contact stress was programmed to increase by 1 MPa every 10 minutes. The seizure load was judged when the torque raised more than 50 N or the oil temperature raised more than 150 °C. The contents of PTFE in PEEK composite were in the range of 0-6 mass%. A Babbitt alloy (3 mass% of Cu, 7 mass% of Sb, and balance of Sn) was selected as the reference. The conditions for TOF-SIMS analysis are given in Table 2.

3. Results and discussion

Island-like particles of PTFE were found on a virgin PEEK-PTFE surface by an optical microscope and an EPMA analysis. However, chemical mapping by TOF-SIMS indicates that PTFE spreads as thin film over PEEK surface (Figure 2). The results can be explained by surface sensitivity of the analysis. The EPMA shows average contents on the surfaces of 1 µm depth, while TOF-SIMS detects the uppermost surfaces of 1-2 nm9). Results of the tribo-test are summarized in Figure 3. The conventional material, Babbitt alloy, showed the seizure pressure of 8-9 MPa.
PEEK provided higher seizure pressure, ranging 9-11 MPa. The blending of PTFE, even at the concentration of 0.5 mass%, considerably improved the load-carrying capacity of PEEK. Seizure was not observed at 12 MPa of the contact pressure that is the maximum load of the tribo-tester. It has been reported that 7.5-20 mass% of PTFE is required for improvement in the tribological properties of PEEK under dry conditions\(^3\)-\(^8\). Much lower concentrations of PTFE are enough to prevent seizure under the lubricated conditions.

We were much interested in the role of PTFE on the load-carrying capacity of PEEK from view points of surface chemistry. Chemical mapping of the rubbed PEEK surfaces indicates that contents and distribution of PTFE were changed during the tribo-test (Figure 4). The island-like area of PTFE was disappeared after the

### Table 2 Conditions for the TOF-SIMS analysis

<table>
<thead>
<tr>
<th>Instrument</th>
<th>Physical Electronics TFS-2100 (TRIFT-II)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Resolution at m/z 26.98 (Al)</td>
<td>M/(\Delta M &gt; 5000)</td>
</tr>
<tr>
<td>Primary ion</td>
<td>bunched (\text{Ga}^+)</td>
</tr>
<tr>
<td>Impact energy, keV</td>
<td>15</td>
</tr>
<tr>
<td>Ion current for one pulse, nA</td>
<td>2</td>
</tr>
<tr>
<td>Pulse width, ns</td>
<td>7</td>
</tr>
<tr>
<td>Bunching intervals, ps</td>
<td>960</td>
</tr>
<tr>
<td>Pulse frequency, kHz</td>
<td>11</td>
</tr>
<tr>
<td>The measured mass range, m/z</td>
<td>0.5–3000</td>
</tr>
<tr>
<td>Total ion doses in measurements, ions (\cdot) cm(^2)</td>
<td>(&lt;1\times10^{12})</td>
</tr>
<tr>
<td>Detected secondary ions</td>
<td>positive and negative</td>
</tr>
<tr>
<td>Area of analysis for chemical mapping, (\mu m^2)</td>
<td>180×180</td>
</tr>
</tbody>
</table>
Changes in contents of the counter surface (steel) were also studied by TOF-SIMS. A fragment ion of m/z 147 that attributes to C$_3$F$_5$O$^+$ was found (Figure 5). The results indicate that transfer of PTFE from the polymer surface onto the steel surface took place by rubbing. Oxidation of PTFE was suggested during the tribo-test, because the compounds on steel surface gave the oxygen-containing fragment ions.

The fluorine contents on the steel surfaces were carefully studied by TOF-SIMS analysis. Together with strong intensity of m/z 19 (corresponds to F$^-$), the fragments of m/z 31 (corresponds to CF$^+$, a fragment ion generated from organic fluoride) and m/z 75 (corresponds to FeF$^+$, fragment ion generated from iron fluoride) were found on the rubbed surfaces. The results indicate that the transfer of PTFE took place both physically (as organic fluorides) and chemically (as iron fluorides).

Taking these results into account, the role of PTFE in improving the load-carrying capacity of PEEK can be summarized in Figure 7. PTFE exists as thin film on virgin PEEK surface. The transfer of PTFE onto the counter surfaces took place during rubbing. As the results, both of the rubbing surfaces are covered with PTFE. Since PTFE is well known as solid lubricants, formation of thin film composed of PTFE on the surface is being beneficial to reduce friction. Therefore, the tribological properties were improved.

It is empirically known that the blending of PTFE in PEEK improves the tribological properties. In the present work, we focused on the role of PTFE from viewpoint of surface chemistry. We wish to note that the present fundamental study has been applied to material design to meet practical applications.

<table>
<thead>
<tr>
<th>Optical microscope</th>
<th>EPMA</th>
<th>TOF-SIMS</th>
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</thead>
<tbody>
<tr>
<td>Optical micrograph</td>
<td>Chemical image of F</td>
<td>Chemical image of CF$^+$ (m/z 30.99)</td>
</tr>
</tbody>
</table>

Figure 2 Morphology and chemical images of virgin PEEK-PTFE (6 mass%) surface

Figure 3 Load-carrying capacity of materials

Figure 4 Chemical images of m/z 31 (CF$^+$) on rubbed PEEK-PTFE (6 mass%) surface
4. Conclusions

Addition of PTFE in PEEK remarkably improved the load-carrying capacity of PEEK-based composites. TOF-SIMS analysis of the rubbed surface afforded the chemical contents on uppermost surfaces of wear track. The results indicate that transfer of PTFE onto the counter surface takes place under the tribological conditions. The importance of surface chemistry in designing new tribo-materials was suggested.
5. Acknowledgement

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6. References


7. Nomenclature

PEEK poly(ether-ether-ketone)
PTFE poly(tetrafluoroethylene)
TOF-SIMS time of flight secondary ion mass spectroscopy
EPMA electron probe micro analysis
m/z mass to charge ratio